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Novel Calix[4]resorcinarene Glycosides

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Abstract: Calix[4]resorcinarene glycosides may be prepared using mild conditions by condensing the corresponding glycosidic aldehyde with resorcinol in the presence of a Lewis acid. The protected glycosyl moieties confer enhanced solubility in a number of organic solvents whilst deprotection provides resorcinarenes which are water soluble. © 1997 Elsevier Science Ltd.

A wealth of knowledge has been amassed concerning calix[n]arenes and the related calix[n]resorcinarenes (resorcinarenes), a recent review detailing the latter's development as useful supramolecular frameworks.^{1,2} Resorcinarenes are attractive for investigation as they are prepared with remarkable ease from readily-available materials. This letter describes the synthesis of glycosidic resorcinarenes as exemplified by compounds (**1a-c**).



The aforementioned review outlines syntheses of resorcinarenes which have been profitably used by Högberg, Cram and Reinhoudt.^{2,3} However, the conditions generally required (alcoholic hydrochloric acid) do not tolerate much variation in functionality of the starting materials. The reviewers note a report in which a simple resorcinarene was prepared using Lewis acid catalysts:⁴ this cached report provided the incentive to explore the use of Lewis acids in the preparation of resorcinarenes from acid-sensitive precursors.

Our efforts have been directed towards the synthesis of resorcinarenes (1a-c) from glycosidic aldehydes (2a-c). The advantages of attaching a glycosidic moiety to the resorcinarene framework are manyfold: the protected carbohydrate confers improved solubility in organic solvents and also aids in the separation of conformational isomers.^{3,4} For example, condensation of aldehyde (2a) with resorcinol in the presence of aluminium chloride gave a resorcinarene, soluble in a wide variety of organic solvents.⁵ Acylation and chromatography enabled the isolation of a major (49%) and a minor (17%) isomer, deduced to be resorcinarenes by analogy with the ¹H NMR spectrum of known compound (4).³ The major product is assigned to be the ctt (chair) conformer and the minor the ccc (boat) conformer. MALDI-TOF mass spectral analysis of both butyrated isomers gave a molecular ion at m/z 2736.

Deacylation of the major isomer with MeOH/H₂O/Et₃N (the product of deacylation is water-soluble), followed by hydrolysis and rebutyration yielded derivative (5), also accessible from a known resorcinarene aglycone.⁶ This enables the positive assignment of structure (6) to the major butyrated resorcinarene glycoside and, hence, structure (1a) to the condensation product from reaction of glycoside (2a) with resorcinol in the presence of aluminium chloride.



As stated above, the carbohydrate moieties confer differing solubility in aqueous or nonaqueous media, depending on their state of protection. Glycosylated calix[4]arenes have been previously reported.⁷ We are currently investigating the use of resorcinarene glycosides as chiral receptors and in the synthesis of novel supramolecular assemblies.

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